

## Review Heat Capacity of Solids

Experimental Facts

- 1) Molar heat capacity is  $3R \sim 24.9 \text{ J K}^{-1}$
- 2) Some exceptions: Diamond  $C_V = 6.1 \text{ J K}^{-1}$
- 3) As  $T \rightarrow 0$   $C_V \rightarrow 0$ , required by the 3rd Law of Thermodynamics
- 4) For vibrations  $C_V \propto T^3$  as  $T \rightarrow 0$  (not explained by Dulong + Petit Rules)
- 5)  $C_V$  of KE of conduction electrons  $C_V \propto T$  have to treat the electrons quantum mechanically to explain.

Calculating Heat Capacity

- 1)  $3N$  degrees of freedom  $\rightarrow 3N$  simple harmonic oscillators
- 2) Quantized energy of one oscillator  $E_n = (n + \frac{1}{2})\hbar\omega$
- 3) Take one oscillator as the system, the rest as the heat bath.
- 4) Derived  $\bar{E} = \frac{\hbar\omega}{e^{\hbar\omega/kT} - 1}$

5) Classical limit  $\hbar \rightarrow 0$   $\bar{E} \rightarrow kT$   $\bar{E} \rightarrow 3N\bar{E} = 3NkT$

6) In classical limit  $C_V = \left(\frac{\partial \bar{E}}{\partial T}\right)_V = 3Nk$  the Dulong - Petit rule

7) Average vibrational energy of the solid

$$\bar{E} = \sum_{\alpha=1}^{3N} \frac{\hbar\omega_\alpha}{(e^{\hbar\omega_\alpha/kT} - 1)}$$

8)  $C_V = -\left(\frac{\partial \bar{E}}{\partial T}\right)_V = k \sum_{\alpha=1}^{3N} \left(\frac{k\omega_\alpha}{kT}\right)^2 \frac{e^{k\omega_\alpha/kT}}{(e^{k\omega_\alpha/kT} - 1)^2}$

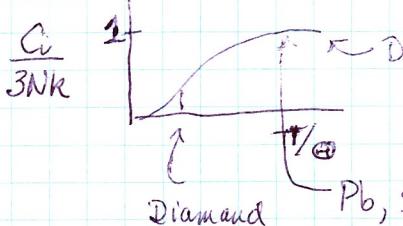
9)  $\omega_\alpha$  - depends on intermolecular forces

10) Atmospheric pressure gives different  $C_p$   
Exact expression in Madel (compressibility, etc.)

11) Einstein's model treated all the  $\omega_\alpha$ 's the same

$$\Theta_E = \frac{\hbar\omega_E}{k} \quad \text{dimension of temperature}$$

$$C_V = 3Nk \left(\frac{\Theta_E}{T}\right)^2 \frac{e^{\Theta_E/T}}{(e^{\Theta_E/T} - 1)^2} \quad k \text{ in } \text{J K}^{-1}$$



$\Theta_E \gg$  Room Temperature  
1300K  
for diamond  $\Theta_E \rightarrow C_V \sim 6.1 \text{ J K}^{-1}$

But Einstein's Model doesn't explain  $C_V \propto T^3$  at low temperatures  
Debye's Model Does! ...

## Calculating $\omega_E$

## Debye Model

You can get  $\omega_E$  from the elastic properties of solids

Definition of Young's Modulus  $Y = \frac{\text{Stress}}{\text{Strain}}$

Treat the lattice as cubic.  
Each atom ~~shares~~ is shared by 8 cubes.

Strain =  $\frac{\epsilon}{a}$      $\epsilon$  = displacement  
               $a$  = interatomic distance

Stress =  $\frac{m\omega_E^2 \epsilon}{a^2}$      $m$  = mass per atom

$$Y = \frac{\epsilon/a}{\text{stress}} = \frac{m\omega_E^2 \epsilon/a}{a^2} = \frac{a}{m\omega_E^2}$$

$$\omega_E = \sqrt{\frac{a}{mY}}$$

$\omega_E = \frac{\hbar\omega_E}{k}$  can get from fitting experimental data

$$\rho = \frac{m}{a^3} = \text{density} \quad M = mN_0 \quad \omega_E = \frac{N_0^{1/3} Y^{1/2}}{\rho^{1/6} M^{1/3}}$$

Diamond is very hard/stiff so  $\omega_E$  is quite large  
that's why Einstein temperature is quite large

## Debye Model

The Debye model is successful low temperatures. It works best for small normal mode frequencies when  $KT \ll$  small.  
Treat the solid as a homogeneous media. Neglect angular frequencies. We can treat the oscillations as elastic waves and use the wave equation:

$$\frac{1}{V^2} \frac{\partial^2 \Psi}{\partial t^2} - \nabla^2 \Psi = 0 \quad \Psi = \text{displacement of an atom in the solid}$$

$V$  phase velocity of the wave

The wave equation predicts either a standing or traveling wave.

Can separate  $\Psi = f(t) \phi(x, y, z)$

where  $f(t) = A \cos \omega t + B \sin \omega t$

## Lecture 14 continued... Density of States of waves in Lattice (Debye model)

$$\frac{\partial^2 \Psi}{\partial t^2} = -\omega^2 \Psi$$

for standing waves

$$-\frac{\omega^2}{V^2} \Psi - \nabla^2 \Psi = 0$$

$$(\nabla^2 + \frac{\omega^2}{V^2}) \Psi = 0$$

$$\nabla^2 \phi f(t) + \frac{\omega^2}{V^2} f(k) \phi = 0$$

the time independent part

$$\frac{\omega}{V} = k$$

$$\nabla^2 \phi + k^2 \phi = 0$$

Cube in cartesian coordinates  
the variables will be separable

$$\Psi = X(x) Y(y) Z(z)$$

Substitute to get

$$\frac{X''}{X} + \frac{Y''}{Y} + \frac{Z''}{Z} = -k^2$$

↗ only depends on x   ↗ only depends on y   ↗ only depends on z  
 $k_1^2 + k_2^2 + k_3^2 = +k^2$

$$\rightarrow \frac{X''}{X} = -k_1^2 \quad \frac{Y''}{Y} = -k_2^2 \quad \frac{Z''}{Z} = -k_3^2$$

The solution is

$$X = A_1 \cos(k_1 x) + B_1 \sin(k_1 x)$$

$$Y = A_2 \cos(k_2 y) + B_2 \sin(k_2 y)$$

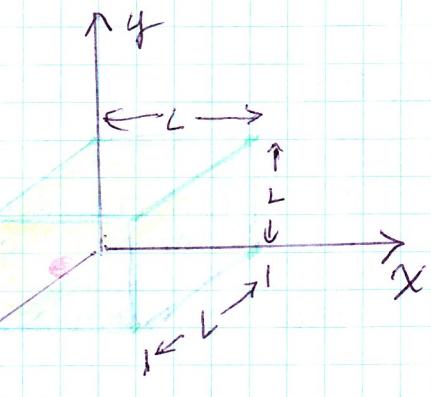
$$Z = A_3 \cos(k_3 z) + B_3 \sin(k_3 z)$$

Subject to the boundary conditions

$$X(0) = X(L) = 0 \quad \therefore k_1 x = n_1 \pi \quad \rightarrow k_1 = \frac{n_1 \pi}{L} \quad n_1 = 0, 1, 2, 3, \dots$$

The normal modes for the elastic waves of a solid

$$n_1 = \frac{k_1 L}{\pi} \quad n_2 = \frac{k_2 L}{\pi} \quad n_3 = \frac{k_3 L}{\pi}$$



It is a cubic lattice and all allowed modes  $n_1, n_2, n_3$  will lie on the cubic lattice. Every point on the lattice corresponds to a normal mode.

There is only 1 mode per unit cell  
the others are shared

Total Volume of unit cell  $(\frac{L}{\lambda})^3$  since  $k_i = (\frac{\pi}{L})n_i$ , etc.

For most of the modes  $n_1, n_2, n_3$  are quite large

There is one allowed mode per unit cell.

What is the number of allowed unit vectors in a spherical shell?

First, just consider one octant

because  $n_1 > 0 \quad n_2 > 0 \quad n_3 > 0$

Want the volume of a spherical shell in that positive octant

$$V = \frac{1}{8} (4\pi k^2 dk)$$

The number of normal modes whose magnitude of  $\vec{k}$  vector  $\vec{k}$  is between  $\vec{k}$  and  $\vec{k} + dk$

$$\text{number of modes} = \frac{1}{8} \frac{(4\pi k^2 dk)}{(\pi/L)^3}$$

$dn = \frac{L^3 k^2 dk}{2\pi^2}$  is the number of modes whose magnitude  $|\vec{k}|$  is between  $|\vec{k}|$  and  $|\vec{k} + dk|$

$$|\vec{k}| = \frac{\omega}{v} \quad \text{when } \omega = v k \quad dw = v dk \quad \frac{dw}{dk} = v \quad \text{no dispersion}$$

$$dn = \frac{V}{8\pi^2} \left( \frac{\omega}{v} \right)^2 \left( \frac{dk}{dw} \right) dw = \frac{V}{28\pi} \frac{\omega^2 dw}{v^3}$$

One more complication: There are two independent transverse modes for every  $\vec{k}$  vector, and one longitudinal mode.  
The expression is more complicated because the velocity is different for transverse and longitudinal components.

$$dn = \frac{V}{28\pi^2} \omega^2 dw \left[ \frac{1}{v_L^3} + \frac{2}{v_T^2} \right]$$

Since there's no dispersion, then can treat as an average velocity  $\frac{3}{\sqrt{3}}$

$$dn = \frac{V}{8\pi^2} \omega^2 dw \frac{3}{v^3}$$

for light,  $v=c$  in vacuum.