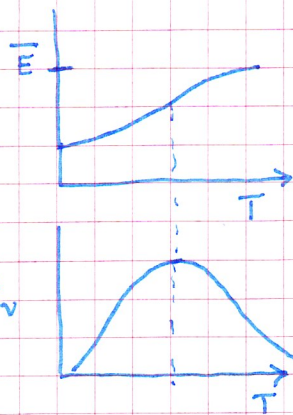


I.
a)



$$\bar{E} = \frac{E_1 + E_2 + E_3}{3} \quad \bar{E} = NE, \quad @ T = 0K$$

$$@ T \rightarrow \infty$$

$$C_v = \frac{\partial \bar{E}}{\partial T}$$

b) Calculate exact expressions

$$Z_1 = \text{partition function for the subsystem} = \sum_r e^{-E_r/KT} = e^{-E_1/KT} + e^{-E_2/KT} + e^{-E_3/KT}$$

$$p_r = \frac{1}{Z_1} e^{-E_r/KT}$$

$$\bar{E}_1 = \sum_r p_r E_r = \frac{E_1 e^{-E_1/KT} + E_2 e^{-E_2/KT} + E_3 e^{-E_3/KT}}{Z_1}$$

$$= \frac{e^{-E_1/KT} (E_1 + E_2 e^{-(E_2-E_1)/KT} + E_3 e^{-(E_3-E_1)/KT})}{e^{-E_1/KT} (1 + e^{-(E_2-E_1)/KT} + e^{-(E_3-E_1)/KT})}$$

$$\bar{E} = N \bar{E}_1 = \frac{N (E_1 + e^{-E_1/KT} + e^{-E_2/KT})}{(1 + e^{-E_1/KT} + e^{-E_2/KT})} \quad \epsilon_1 = E_2 - E_1, \quad \epsilon_2 = E_3 - E_1$$

$$C_v = \frac{\partial \bar{E}}{\partial T} = \frac{N (E_1 + e^{-E_1/KT} + e^{-E_2/KT}) (-1) [e^{-E_1/KT} (-\frac{E_1}{K}) (-\frac{1}{T^2}) + e^{-E_2/KT} (-\frac{E_2}{K}) (-\frac{1}{T^2})]}{(1 + e^{-E_1/KT} + e^{-E_2/KT})^2}$$

$$+ \frac{N [e^{-E_1/KT} (-E_1/K) (-1/T^2) + e^{-E_2/KT} (-E_2/K) (-1/T^2)]}{(1 + e^{-E_1/KT} + e^{-E_2/KT})}$$

$$= \frac{N}{KT^2} (1 + e^{-E_1/KT} + e^{-E_2/KT}) (E_1 e^{-E_1/KT} + E_2 e^{-E_2/KT}) - \frac{N (E_1 + E_2 e^{-E_2/KT})}{(1 + e^{-E_1/KT} + e^{-E_2/KT})^2}$$

$$C_v = \frac{N}{KT^2} \frac{(E_1 e^{-E_1/KT} + E_2 e^{-E_2/KT}) [(1 - E_1) + (1 - E_2) e^{-E_1/KT} + (1 - E_2) e^{-E_2/KT}]}{(1 + e^{-E_1/KT} + e^{-E_2/KT})^2}$$

easier to just write:

$$\bar{E} = \frac{N \sum_r E_r e^{-E_r/KT}}{\sum_r e^{-E_r/KT}} = \frac{N \sum E_r e^{-E_r/KT}}{Z} \quad \frac{\partial \bar{E}}{\partial T} = -\frac{1}{Z^2} \left(\sum E_r^2 e^{-E_r/KT} \right) = \frac{-E_r}{Z^2 KT^2}$$

$$C_v = \frac{\partial \bar{E}}{\partial T} = \frac{N}{KT^2} \frac{\sum E_r^2 e^{-E_r/KT}}{\sum e^{-E_r/KT}} + \frac{N}{Z} \sum E_r e^{-E_r/KT} \left(\frac{-E_r}{K} \right) \left(\frac{-1}{T^2} \right) = \frac{N}{KT^2} \frac{\sum E_r^2 e^{-E_r/KT}}{\sum e^{-E_r/KT}}$$

II

P_1	T	P_2	T
V_1	N	V_2	N

P	V	T	$2N$
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$$P_1 V_1 = NkT$$

$$P_2 V_2 = NkT$$

$$V_1 + V_2 = V_0 = \frac{2NkT}{P} \rightarrow \frac{2}{P} = \frac{1}{P_1} + \frac{1}{P_2} = \frac{P_1 + P_2}{P_1 P_2} \rightarrow P = \frac{2P_1 P_2}{P_1 + P_2}$$

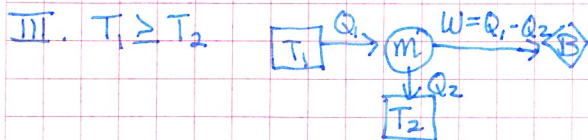
how much has entropy changed?

$$\Delta S = \Delta S_1 + \Delta S_2 \quad dE = dQ + dW = 0 \rightarrow dQ = PdV$$

$$\Delta S_1 = \int \frac{dQ}{T} = \int \frac{PdV}{T} = - \int \frac{VdP}{T} = -Nk \int_{P_1}^{P_f} \frac{dP}{P} = -Nk \ln\left(\frac{P_f}{P_1}\right) \quad \Delta S_2 = -Nk \ln\left(\frac{P_f}{P_2}\right)$$

$$\Delta S = -Nk \ln\left(\frac{P_f^2}{P_1 P_2}\right) = -Nk \ln\left[\frac{4(P_1 P_2)^2}{P_1 P_2 (P_1 + P_2)^2}\right] = Nk \ln\left[\frac{(P_1 + P_2)^2}{4P_1 P_2}\right]$$

Since $(P_1 + P_2)^2 > 4P_1 P_2$ \therefore total entropy change is positive.



Let's assume $\Delta S_m = 0$ $\Delta S_B = 0$ entropy change is in the reservoir

$$Q_1 = C(T_1 - T_f) \quad Q_2 = C(T_f - T_2) \quad W = Q_1 - Q_2 = C(T_1 + T_2 - 2T_f)$$

reversible: $\Delta S_{TOT} \geq 0 \Rightarrow C \ln \frac{T_f}{T_1} + C \ln \frac{T_f}{T_2} \geq 0 \rightarrow \ln \frac{T_f^2}{T_1 T_2} \geq 0 \quad T_f \geq \sqrt{T_1 T_2}$

decreases increases

as T_f increases, the work available increases

$$W \leq C(T_1 + T_2 - 2\sqrt{T_1 T_2}) \quad W_{max} = C(\sqrt{T_1} - \sqrt{T_2})^2$$

What is the maximum efficiency $\eta_{max} = \frac{W_{max}}{Q_1} = \frac{C(\sqrt{T_1} - \sqrt{T_2})^2}{C(T_1 - \sqrt{T_1 T_2})} = \left(\frac{\sqrt{T_1} - \sqrt{T_2}}{\sqrt{T_1}}\right)$
 (because heat reservoirs don't have infinite capacities $\eta_c = \frac{T_1 - T_2}{T_1} = .64$)

IV $\mu = 10^{-26} \text{ Am}^2 \quad p_+ = \frac{e^x}{e^x + e^{-x}} = .8 \rightarrow 2e^x = 8e^{-x} \rightarrow e^{2x} = 4 \rightarrow x = \frac{1}{2} \ln 4$

$$x = \frac{\mu B}{kT} = \frac{1}{2} \ln 4 \rightarrow B = \left(\frac{kT}{\mu}\right) \frac{1}{2} \ln 4 = \frac{1.38 \cdot 10^{-23} \text{ JK}^{-1} (10^{-2} \text{ K}) \ln 4}{2(10^{-26} \text{ Am}^2)}$$

$$B = \frac{(1.38)(10)(\ln 4)}{2} \text{ Tesla}$$